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CHAPTER 6: THERMOCHEMISTRY 168 6.20 Strategy: The work done in gas expansion is equal to the product of the external, opposing pressure and the change in volume.  $w = -P_{\text{ext}}\Delta V$  We assume that the volume of liquid water is zero compared to that of steam.

### CHAPTER 6 THERMOCHEMISTRY

Chapter 6 Equilibrium Chemistry 211 the reaction we monitor the mass of  $\text{Ca}^{2+}$  remaining in solution and the mass of  $\text{CaCO}_3$  that precipitates, the result looks something like Figure 6.1. At the start of the reaction the mass of  $\text{Ca}^{2+}$  decreases and the mass of  $\text{CaCO}_3$  increases. Eventually the reaction reaches a point after which there

### Chapter 6

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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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Chapter 6 Chemistry in Biology Neutrons and protons are located at the center of the atom. ... 6.3 Water and Solutions Chapter 6. Chemistry in Biology The measure of concentration of  $\text{H}^+$  in a solution is called pH. Acidic solutions have pH values lower than 7.

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