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An acid is a molecule or ion capable of donating a hydron (proton or hydrogen ion  $H^+$ ), or, alternatively, capable of forming a covalent bond with an electron pair (a Lewis acid).. The first category of acids is the proton donors or Brønsted acids. In the special case of aqueous solutions, proton donors form the hydronium ion  $H_3O^+$  and are known as Arrhenius acids.

### **Acid - Wikipedia**

In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations and negatively charged ions called anions. These can be simple ions such as the sodium ( $Na^+$ ) and chloride ( $Cl^-$ ) in sodium chloride, or polyatomic species such as the ammonium ...

### **Ionic compound - Wikipedia**

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