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Determination of Fe<sup>2+</sup> by Redox Titration with MnO<sub>4</sub>. Acid-Base Titration Lab. Fresh ... Englewood Cliffs, NJ. Prentice Hall. Pages 613-616.

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### **AP Chemistry Laboratory #17: Determination of Unknown Acid**

In this titration. and phenolphthalein indicator is added. 0 mL HCl. GHS Honors Chemistry Acid-Base Titration Lab Name: \_\_\_\_\_ Date: \_\_\_\_\_ Block: \_\_\_\_\_ standardized solution of NaOH Phenolphthalein indicator o Unknown solution of HCl o o -3- .

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The pH of these titrations can be monitored using a pH meter, an indicator, and a conductivity meter as a function of volume, and this data can be saved to an electronic lab book for later analysis. A smaller set of potentiometric titrations can also be performed.

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Demonstration: Titration of an Acid with a base. Lab. Acids, Bases and pH. Journal Article. Conceptual and Procedural Knowledge of Acid-Base Behavior of Substances, ... Prentice Hall Lab Manual . Study Guide

**Science Topic/Unit: Chemistry/Chemistry A – Acids and**

Titration Page 3 of 5 Once the exact molarity of the NaOH solution is known, the base can be used to determine the amount of KHP or any other acid present in a known mass of an impure sample. Example: The percentage determination of an acid in an unknown sample. 0.5366 g of an KHP sample of unknown purity was massed.

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