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Sulfur hexafluoride (SF₆) is an inorganic, colorless, odorless, non-flammable, extremely potent greenhouse gas, and an excellent electrical insulator. SF₆ has an octahedral geometry, consisting of six fluorine atoms attached to a central sulfur atom. It is a hypervalent molecule. Typical for a nonpolar gas, it is poorly soluble in water but quite soluble in nonpolar organic solvents.

Sulfur hexafluoride - Wikipedia

At room temperature, sodium hydroxide is a white crystalline odorless solid that absorbs moisture from the air. It is a manufactured substance.

Sodium hydroxide | NaOH - PubChem

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